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Vol. 15 No. 1 (2024)



DOI: <https://doi.org/10.21776/jrm.v15i1>

Published: 2024-05-15

Articles

PENGARUH WAKTU SOLUTION TREATMENT DAN AGING TERHADAP KEKERASAN DAN KEKUATAN TARIK ALUMINIUM PADUAN AA 7075 – T6

Amira Naafila, Anindito Purnowidodo, Putu Hadi Setyarini

1-12

PDF

STUDY OF ADDITIONAL FIN TO INCREASE EFFICIENCY OF SUPERHEATER AT HEAT RECOVERY STEAM GENERATOR

Muhammad Agung Bramantya

13-26

PDF

PEMANFAATAN ZEOLIT NAA DARI KAOLIN SEBAGAI ADSORBEN LOGAM BESI (FE) PADA LIMBAH CAIR LABORATORIUM

Ulul Khairi Zuryati, Ruru Honiar, Fahrurazi Fahrurazi

27-36

[PDF](#)**ANALISIS KARAKTERISTIK SUDUT PENYEMPROTAN BIODIESEL PALM OIL MELALUI PENAMBAHAN VARIASI KONSENTRASI METANOL**

Muhammad Suryaningrat, Mega Nur Sasongko, Nurkholis Hamidi, Nurhadi Saputro
37-49

[PDF](#)**PENGARUH PENGUJIAN KEKERASAN DAN PENGUJIAN TEKAN TERHADAP KARAKTERISTIK GAYA TUMBUKAN PADA CANGKANG KENARI (CANARIUM INDICUM L)**

Leslie S. Loppies, Alexander A. Patty, Berthy Pelasula, Roy R. Lekatompessy, Nevada Mario Nanulaitta
51-60

[PDF](#)**PEMILIHAN JENIS AIRFOIL MOHINDER UNNES MENGGUNAKAN METODE ANALYTICAL HIERARCHY PROCESS**

Dony Al-Janan, Yoga Gusda
61-71

[PDF](#)**PENGARUH JUMLAH SUDU TERHADAP KINERJA TURBIN SAVONIUS TIPE DRAG PADA ALIRAN AIR DALAM PIPA**

Agus Jamaldi, Arif Hidayat Purwono, Deni Andriyansyah, Emanuel Budi Raharjo
73-82

[PDF](#)**PENGARUH PENAMBAHAN ADJUSTER PADA KOMPOR BRIKET TERHADAP JARAK PEMBAKARAN DAN LAJU PERPINDAHAN PANAS KONDUKSI**

Jusuf Haurissa, Helen Riupassa, Hendry Nanlohy, Suyatno Suyatno
83-92

[PDF](#)**PENGARUH VARIASI SUDUT DAN JARAK PENEMBAKAN SANDBLASTING PADA PROSES PRETREATMENT TERHADAP KUALITAS PELAPISAN CAT DAN SIFAT KOROSI PADA BAJA KARBON RENDAH**

Indra Adi Saputra, Teguh Dwi Widodo, Lilis Yuliati
93-102

[PDF](#)**THE EFFECT OF THERMAL RESISTANCE OF CEILING LAYER ON THE PERFORMANCE OF AN AIR**

CONDITIONER AND ROOM TEMPERATURE DISTRIBUTION

Jeri Tangalajuk Siang, Daud Patabang
103-114

 PDF**PENINGKATAN PRODUKSI BAHAN BAKAR HIDROGEN DENGAN BANTUAN NATURAL SURFACTANT PADA PROSES WATER SPLIT**

Purnami Purnami, Fransisca Gayuh Utami Dewi, ING Wardana, Mega Nur Sasongko, M Umar Yusuf, Willy Satrio Nugroho
115-124

 PDF**Kekuatan Bending dan Tarik Komposit Berpenguat Serat Eceng Gondok/Tebu Bermatrik Epoxy**

Rahmat Doni Widodo, Herry Sutanto, Deni Fajar Fitriana, Rusiyanto Rusiyanto, Febri Budi Darsono
125-138

 PDF**POTENSI LIMBAH CANGKANG KERANG SEBAGAI KATALIS HETEROGEN UNTUK PEMBUATAN BIODIESEL**

Abdi Hanra Sebayang, Muhammad Anhar Pulungan, Sihar Siahaan, Siti Maretia Benu, Husin Ibrahim, Munawar Alfansury Siregar, Arridina Susan Silitonga
139-147

 PDF**OPTIMASI PROSES TURNING PADA AA 6061 DENGAN METODE MINIMUM QUANTITY LUBRICATION**

Mohammad Anshori, Achmad As'ad Sonief, Putu Hadi Setyarini
149-156

 PDF**ANALYSIS OF COCONUT FIBER REINFORCED COMPOSITES WITH HOT PRESS TECHNIQUES**

Seno Darmanto, Alvin Dio Nugroho, Nur Kholis Fathurrohman , Imam Saputra , Muhammad Kusni, Muhammad Akhsin Mulfikhun
157-166

 PDF**PEMBUATAN DAN PENGUJIAN SIFAT MEKANIK WELDING SPESIMEN LOGAM DISSIMILAR AISI 1015 DENGAN AISI 304L MENGGUNAKAN PROSES GMAW**

Riswanda Riswanda, Harlian Kadir, Albert Saragih

167-175

[!\[\]\(bd1a142de767a21e5362c595f844a4ff_img.jpg\) PDF](#)**PENGEMBANGAN PRODUK CASING ALAT KESEHATAN PEMANTAUAN INFUS PINTAR (MIFUS) DENGAN METODE RAPID TOOLING**

Riona Ihsan Media, Roni Kusnowo, Yogi Muldani Hendrawan, Hafez Trimukti Ali Musa

177-193

[!\[\]\(830769b31eeeaca920791081939ff8ba_img.jpg\) PDF](#)**DESIGN OF NATURAL FIBER POWDER MACHINE**

Ali Attamimi, Alfan Ekajati Latief, Zidane Nassem, Riski Gunarto Ramadan

195-207

[!\[\]\(47734e4656765d20df4fdbd5b7aff048_img.jpg\) PDF](#)**PEMBUATAN BETWEEN CENTRE TEST BAR MESIN BUBUT UNIVERSAL**

Risky Ayu Febriani, Novi Saksono, Addonis Candra, Herma Budi Harja, Rivaldy Muhammad Fahlevi

209-219

[!\[\]\(41aea2746216b27a6939d696d8e035da_img.jpg\) PDF](#)**STUDI NUMERIK KINERJA PENDINGIN COOLANT HYBRID-NANOFUID AL2O3-TIO2 PADA RADIATOR**

Najmul Hidayat, Sudarmadji Sudarmadji

221-231

[!\[\]\(179f167ede0522ebb4ea025b3ad78ca7_img.jpg\) PDF](#)**ANALYSIS OF COGGING TORQUE REDUCTION FROM DESIGN COMPUTATIONAL PERMANENT MAGNET SYNCHRONOUS MOTOR WITH TAGUCHI METHOD**

Fitri Wahyuni, James Julian, Ferdyanto Ferdyanto, Ade Fikri Fauzi

233-245

[!\[\]\(5ddb2a112276baa148775929432349f9_img.jpg\) PDF](#)**ANALISIS MODIFIKASI DESAIN REAKTOR PADA RANCANG BANGUN ALAT PIROLISIS DAN PENGUJIAN NILAI KALOR UNTUK PLASTIK PP DAN ABS**

Rudi Purwo Wijayanto, Francois Rubian Alhikam, Iyus Hendrawan

247-256

[!\[\]\(135faf555a2da147cc447132eda26e60_img.jpg\) PDF](#)**ANALISIS KARAKTERISTIK ALIRAN MELALUI PENAMPANG PERSEGI PANJANG MENGGUNAKAN MODEL TURBULEN LARGE EDDY SIMULATION (LES)**

La Ode Ahmad Barata, Samhuddin Samhuddin

257-272

 PDF

ANALISIS KONSUMSI ENERGI LISTRIK PADA PEMBANGUNAN NUWSP BIYONGA KABUPATEN GORONTALO

Rahmad Hidayat Boli, Rifaldo Pido, Mohamad Rifal Arbie, Wawan Rauf

273-281

 PDF

MODEL 3D IMPLAN LUTUT FEMUR DARI REKONSTRUKSI TULANG LUTUT DENGAN METODE REVERSE ENGINEERING BERBANTUAN PEMINDAI CT-SCAN

Suryadiwansa Harun, Helmi Ismunandar, Yanuar Burhanuddin, Satrio Darma Supriyadi

283-296

 PDF

CARBON SIZE AND TEMPERATURE EFFECTS TO JIS S45C CARBURIZED STEEL

I Made Wicaksana Ekaputra, Yustinus Akas Wibisono, Gunawan Dwi Haryadi

297-303

 PDF

PENGARUH KOMPOSISI LAPISAN NI-CR PADA BAJA ASME SA 210 C TERHADAP LAJU EROSI SUHU TINGGI

Slamet Prasetyo Utomo, Djarot B. Darmadi, Teguh Dwi Widodo

305-320

 PDF

EVALUASI KINERJA DIRECT EXPANSION SOLAR-ASSISTED HEAT PUMP WATER HEATER KAPASITAS 30 LITER DENGAN MENGGUNAKAN REFRIGERAN R134A

Teguh Irawan, Rahmat Iman Mainil, Azridjal Aziz

321-327

 PDF

EFFECT OF COMPRESSOR INLET TEMPERATURE ON THERMAL EFFICIENCY ROLLS ROYCE RB211 GAS GENERATOR IN COMBINED CYCLE POWER PLANT

Nanang Yulistio, Muhammad Prihadi Eko Wahyudi, Qoriatul Fitriyah

329-337

 PDF

ANALISIS PENGARUH VARIASI ARAH PENGELASAN HARDFACING TERHADAP KEKUATAN TARIK DAN STRUKTUR MIKRO PADA BAJA ASTM A36

Jatmoko Awali, M. Arzil Maulana, Fikan Mubarok Rohimsyah, Yunita Triana
339-346

 PDF

ANALISIS KINERJA MESIN PEMARUT SAGU

Daniel Parenjen, Suwarjono Suwarjono, Cipto Cipto
347-354

 PDF

STRENGTH CONSIDERATION ON CAR BODY MODIFICATION FOR PANORAMIC TRAIN

Dinny Harnany, Reyhan K. A. Adista, Achmad Syaifudin, Ary Bachtiar Krishna Putra, Singgih Priyambodo
355-366

 PDF

PENGARUH CO-FIRING SERBUK KAYU KEDONDONG TERHADAP PERFORMA DAN EMISI GAS BUANG BRIKET BATU BARA

Edi Nuryanto, Sudarno Sudarno, Yoyok Winardi
367-375

 PDF

PENGARUH OUTSERT TERHADAP ALIRAN DAN PERPINDAHAN KALOR PADA ANNULAR HEAT EXCHANGER TIPE HORIZONTAL

Mustaza Ma'a, Samsul Kamal, Indro Pranoto
377-392

 PDF

ANALYSIS THE EFFECTS OF YTTERBIUM RARE EARTH ON THE CORROSION RATE OF SACRIFICIAL ANODE ALUMINIUM IN SEAWATER ENVIRONMENT AT ROOM TEMPERATURE

Faisal Nurdiansyah, Bima Tegar Pribadi, Tubagus Noor Rohmannudin, Budi Agung Kurniawan
393-410

 PDF

PYROMETALLURGICAL PROCESS FOR ZINC ANALYSIS IN SPHALERITE APPLYING XRD AND XRF

Wiwik Dahani, Riskaviana Kurniawati, Rita Sundari, Irfan Marwanza, Faisal Rachman
411-424

 PDF

MECHANICAL PROPERTIES OF STEEL AISI 1045 VARIATION OF AUSTENITIZATION HOLDING TIME IN THE QUENCHING-TEMPERING PROCESS WITH ICE WATER MEDIA

Ade Yusariarta Putra Parmita, Yogi Mirza Pangestu Utomo, Arie Mifthahul Rakhmat, Muthia Putri Darsini Lubis
425-432

 PDF**EFFECT OF ANNEALING ON MICROSTRUCTURE AND HARDNESS OF FeNiCo ALLOYS SYNTHESIZED BY MECHANICAL ALLOYING**

Sutrisna Sutrisna, Didit Setyo Pamuji, Angger Bagus Prasetyo, Ismail Zulpria Ababil, Ihwanul Aziz
433-442

 PDF**PENGARUH KADAR AIR DI DALAM BRAKE FLUID TERHADAP KARAKTERISTIK GELEMBUNG PADA PROSES PEMANASAN BRAKE FLUID**

Setya Wijayanta, Faris Humami, Helmi Wibowo, Komang Andre Kristiawan, Wildan Surya Lazuardi
443-454

 PDF**ANALISIS PENGGUNAAN SISTEM AKTUATOR PNEUMATIK BERTENAGA SURYA TERHADAP KINERJA MESIN PEMOTONG NANAS**

Rafil Arizona Rafil, M Iqbal, Kurnia Hastuti Kurnia, Sehat Abdi Saragih Sehat, Shandy Kurniadi Shandy
455-472

 PDF**COMPARISON OF THE ACTIVATION OF GIGANTOCHLOA APUS TO INCREASE THE ADSORPTION ABILITY OF MEDICAL LIQUID WASTE**

Putu Hadi Setyarini, Jemmie Iksandy, M. Qashmal Fachrezi, Achmad As'ad Sonief
473-482

 PDF**KARAKTERISTIK MEKANIK DAN BALISTIK KOMPOSIT Al-8Zn-4Mg-20 VOL.% SiC HASIL PROSES SQUEEZE CASTING DAN PERLAKUAN PENUAAN**

Absaralita Sabarati, Panji Adhipura, Bondan T. Sofyan
483-494

 PDF**DURASI PERAWATAN OPTIMUM UNTUK FASILITAS PRODUKSI MINYAK DAN GAS DI KILANG TUBAN EAST JAVA**

Indra Nur Yahya, Dian Mahatwan, Ari Setiawan, Djarot B. Darmadi

495-506

[PDF](#)**JARAK ROTOR YANG OPTIMAL TERHADAP RIPPLE PLATE PADA MESIN RIPPLE MILL UNTUK EFISIENSI HASIL PEMECAH BIJI KELAPA SAWIT CB MODIPALM KAPASITAS 8 TON/JAM**

Agus Harianto, Aspiyansyah Aspiyansyah, Ecy Yedija Faot

507-513

[PDF](#)**ANALISIS PERUBAHAN BENTUK CHASIS TERHADAP BEBAN SEBAGAI DASAR PERANCANGAN MESIN BENDING HIDRAULIC**

Makinun Makinun, Agus Dwi Anggono, Tri Widodo Besar Riyadi

515-524

[PDF](#)**STEAM TURBINE HEAT RATE ANALYSIS BY METHOD PERFORMANCE TESTS ON POWER PLANT IN PT. KSA PALM FACTORY**

Alfan Khusnaini, Aspiyansyah Aspiyansyah, Agus Harianto, Halim Adrianur

525-532

[PDF](#)**UNJUK KERJA MOTOR BAKAR PERTALITE DENGAN PENAMBAHAN BIOETHANOL DARI TETES TEBU**

Ari Purnama, Macrus Afif Romdloni

533-540

[PDF](#)**PENGARUH LOAD CAPACITY LISTRIK TERHADAP EFFISIENSI TURBIN UAP MODEL C6-R8-ER : STUDY KASUS PADA PT. SURYA BORNEO INDUSTRI**

Teddy Tratama, Aspiyansyah Aspiyansyah, Agus Harianto, Kukuh Rahmandika

541-546

[PDF](#)**ANALISIS BENTUK PENAMPANG BULAT, SEGI EMPAT DAN BULAT BERONGGA TERHADAP FATIGUE FAILURE DAN TEGANGAN PADA MATERIAL BAJA ST 60**

Aspiyansyah Aspiyansyah, Adriant Widayat

547-552

[PDF](#)

PENINGKATAN KINERJA MESIN 2 SILINDER DENGAN PENAMBAHAN OCTANE BOOSTER PADA RON 90

Muhammad Arif Hariyadi, Aspiyansyah Aspiyansyah, Agus Harianto, Agus Adjie Riduan
553-560

 PDF

About Rekayasa Mesin

[Focus and Scope](#)

[Editorial Team](#)

[Reviewer Acknowledgement](#)

[Publication Ethics](#)

[Visitor Statistics](#)

[Author Fees](#)

Information for Author

[Author Guidelines](#)

[Template](#)

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[Author Guidelines](#)

[Template](#)

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* [Template](#)

* [Submission](#)

* [Revision](#)

* [Decision](#)

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Index

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[Publication Ethics](#)

[Visitor Statistics](#)

[Author Fees](#)

CARBON SIZE AND TEMPERATURE EFFECTS TO JIS S45C CARBURIZED STEEL

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Abstract

Wear on a metal can be naturally found in components that operate in a friction environment, such as in a transmission system. However, it can be a substantial problem when the wear rate is not within the permitted limit. Surface hardening is a popular way to improve surface wear resistance for metals with various parameters that can be controlled during the process. In this study, pack carburizing was conducted for plain carbon steel JIS S45C. The carburizing process is conducted on the steel by varying the carbon size and temperature during the heating process. The carbon size consisted of mesh sizes of 5 and 15. The heating process was conducted for 2 hours at 800°C and 900°C. The carbon media and catalyst used in this study were coconut charcoal and calcium carbonate (CaCO₃). In addition, the hardness Vickers test was conducted to evaluate the surface hardness. It was found that the hardness of Vickers increased with an increase in heating temperature and mesh size of carbon.

Keywords: Carbon Size, CaCO₃, JIS S45C, Wear Rate.

1. INTRODUCTION

The naming of the JIS S45C steel alloy refers to the standards set by Japanese industrial standards where the steel is categorized as plain carbon steel with a carbon content of 0.45% wt. ^[1-2]. If it is compared to the American Iron and Steel Standard (AISI) standard, JIS S45C is equivalent to AISI 1045. JIS S45C is mainly used for the transmission components such as shafts, gears, clutch pulleys, etc. ^[2]. Due to its working environment, JIS S45C steel must withstand a certain level of wear. Therefore, an effort is needed to develop the wear resistance of this steel alloy. Case hardening by applying particular heat treatment conditions on the metal surface has been applied for particular alloy steel for increasing the wear resistance. It has been reported that a significant hardness has been increased by applying case hardening. Several case hardening methods were divided based on the media used for surface coating, for example, carburizing, nitriding, etc. ^[3-6].

Pack carburizing is an economical and effective method that can increase the hardness of a metal surface. However, not all types of steel can be applied to the pack carburizing process ^[5-8]. For example, steel alloys with low carbon composition make the pack carburizing process less effective because carbon is insufficient to form a martensitic structure ^[5-6]. Conversely, the pack carburizing process on high carbon steel can cause the steel to become very brittle and break easily due to the amount of martensite structure that is too large ^[8]. Based on some literature, most types of medium steel are ideal for heat treatment

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Received on: 2023-02-25

Revised on: 2023-05-26

Accepted on: 2023-09-20

because of the effectiveness of atomic diffusion time on the amount of carbon present during the heating and cooling processes [1-2, 8-14]. Furthermore, the carbon size can also determine the effectiveness of surface hardening for the steel [12].

Negara et al. conducted a pack carburizing study on low-carbon steel types using carbon media derived from bamboo charcoal and a catalyst called barium carbonate (BaCO_3) [6]. The composition of carbon and catalyst used in the furnace is 80% and 20%, respectively. The hot temperature for the pack carburizing process is 950 °C, with a holding time of approximately 4 hours. In addition, the Vickers hardness testing process is carried out to determine the hardness trend formed in the steel. Significant increase in steel hardness and strength on the steel surface after the pack carburizing process.

On the other hand, the toughness value has decreased. Sujita conducted research related to the surface hardening process using the pack carburizing method on AISI 1018 steel [5]. The carbon media comes from finished wood charcoal, and the catalyst media comes from golden snail shell powder. The carburizing process was carried out at several temperature variations with two holding times of 90 and 150 minutes. From the results of the Vickers hardness test, it was found that the increase in hardness occurred in a composition of 80% wood charcoal and 20% golden snail shell powder. This increase can also be seen from the dominance of carbide on the steel surface.

In this study, an investigation regarding the effect of the pack carburizing process on the surface hardness level of JIS S45C carbon steel was investigated. Parameters of time and carbon mesh size were applied to determine the level of hardness on the steel surface. The hardness was examined by the Vickers hardness method following the ASTM E384. In addition, the microstructure of the hardened surface was observed by using an optical microscope.

2. METHODS

The pack carburizing process was carried out in a fully packed condition. The steel metal was placed in a box (packing) made of steel, and then a carbon medium and a catalyst were placed around the box. For the optimal carburization process, the sidelines of the box were sealed entirely to prevent the carbon escaped. Finally, the box was put into the furnace to be heated at a specific carburizing for 2 hours and under two different temperatures of 800 and 900 °C. After the heating process, the carburized specimen was quenched in the water. A typical carburization method in this study can be schematically illustrated in Figure 1.

Prior carburizing process, the specimen of JIS S45C was normalized. After the carburizing process, the Vickers hardness method examined the hardness by following ASTM E384 [15]. The parameters developed in the carburizing process were the heating temperature of 800 and 900 °C, and the carbon mesh size at 5 and 10 mesh. The carbon media comes from coconut charcoal, and the catalyst is CaCO_3 . Microstructure observations on the carburized surface were carried out using an optical microscope to determine the physical impact of the carburizing process.

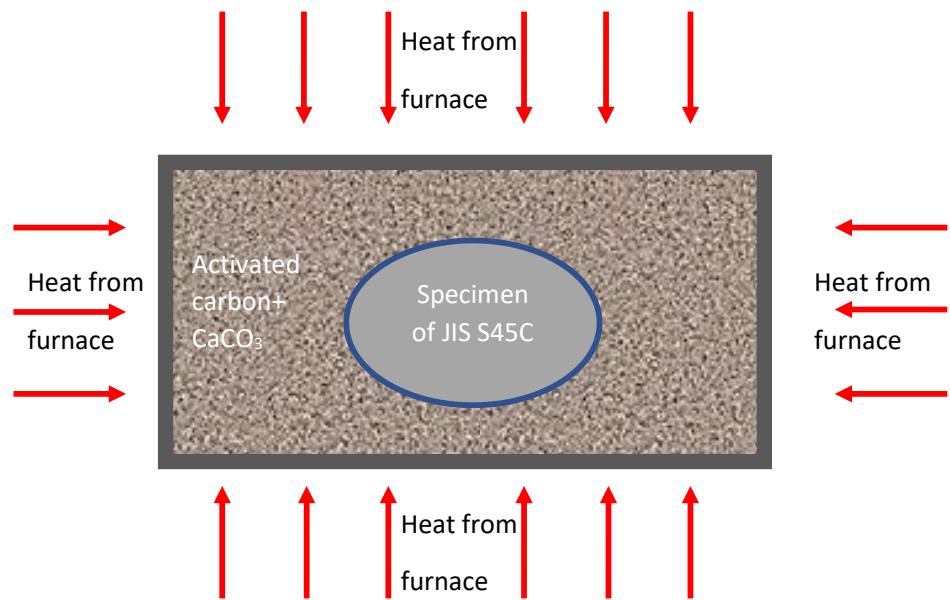


Figure 1. Schematic illustration of carburizing process for JIS S45C.

3. RESULTS AND DISCUSSIONS

Figure 2 shows the hardness Vickers result for JIS S45C as-received (normalized) and carburized steel. The as-received has the lowest hardness of 193 HV since a typical normalized process will recover the residual stress and increase the toughness. Typical normalized steel has a ferrite and pearlite structure, as shown in Figure 3. Ferrite is known to have a characteristic of ductility, while pearlite is a compound of carbon and ferrite, increasing steel's strength. Since the normalized steel is dominated by ferrite, the hardness will be the lowest compared to the other treated steel. After the normalizing process, all steel performs a significant hardness on its surface after being carburized. Higher heating temperatures and mesh carbon size will increase the surface hardness. The highest hardness is obtained at 850 HV when the heating temperature was applied at 900 °C with a mesh carbon size of 15. The higher carbon mesh means a smaller size of carbon.

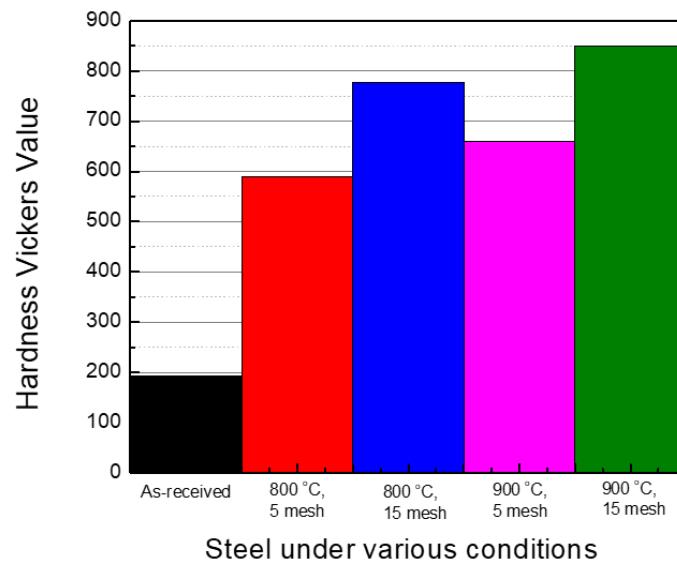


Figure 2. Vickers hardness test result for JIS S45C.

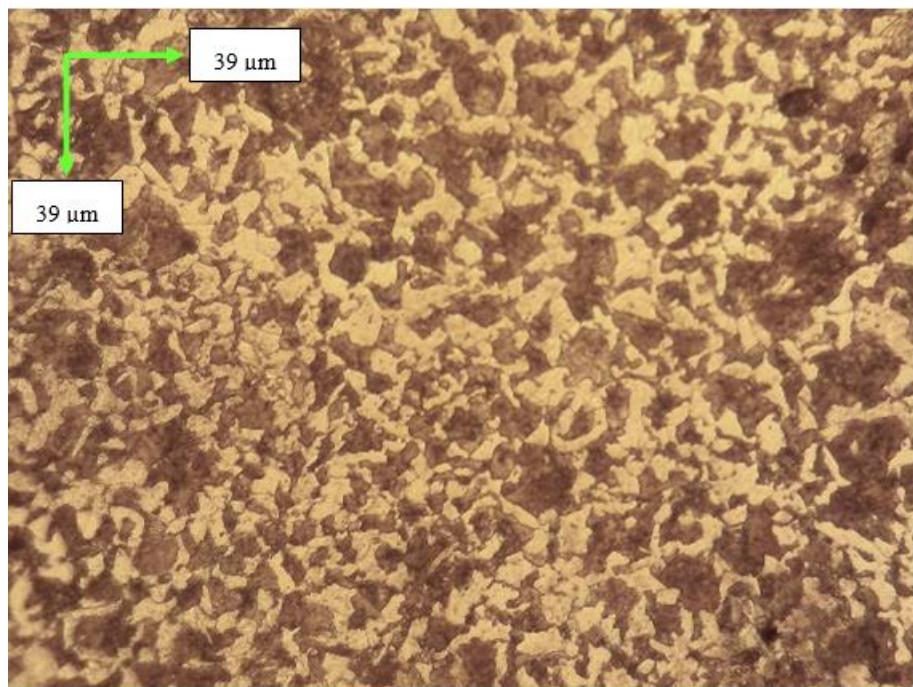


Figure 3. The microstructure for as-received (normalized) JIS S45C.

The heating temperature significantly affects the hardness of JIS S45C since temperature influences the solubility rate of carbon on the steel. The applied temperature in this study was applied an austenized temperature. Hence when the cooling time is applied rapidly, the hardest martensite structure is developed. Figure 2 shows that the level of austenite temperature gives an increase in hardness. The martensite structure will be more massive in quantity since the rate of carbon solubility becomes higher, as shown in Figures 4 and 5. The prediction method for calculating the quantity of microstructural components in steel cooled from the austenitizing temperature using the regression method has been reported by Trzaska [17]. By following this method, the martensite quantity for JIS S45C at 800 °C with five mesh sizes the quantity around 70%, and the for JIS S45C at 900 °C with 15 mesh has the quantity around 80%. Furthermore, the hardness increase can also be produced due to high dislocation density at higher austenite temperatures. Abidah et al. emphasized that the dislocation density due to carbon solubility will be generated with higher and extended holding temperatures [4].

From Figure 2, the hardness increases with the larger mesh or, the smaller carbon size. Silberberg has reported that the chemical reaction rate will increase when the carbon element becomes smaller [16]. The smaller size of carbon caused the comparison of the area with volume to become more significant. With the large surface area, the contact surface between carbon and another element will be generated in large quantities. This large contact surface will increase the hardness. According to Adly et al., the smaller size of carbon can generate a small gap between carbon grains [9]. The small gap will have a small size of oxygen that obstruct the production of CO₂. With the extra quantity of CO₂, the carburization process rate will vary. Leman et al. reported that there was an optimum carbon grain size to produce the CO₂. The optimum carbon grain size will increase the hardness. This preliminary experimental result shows that the highest hardness is obtained at the carbon grain size of 15 mesh.

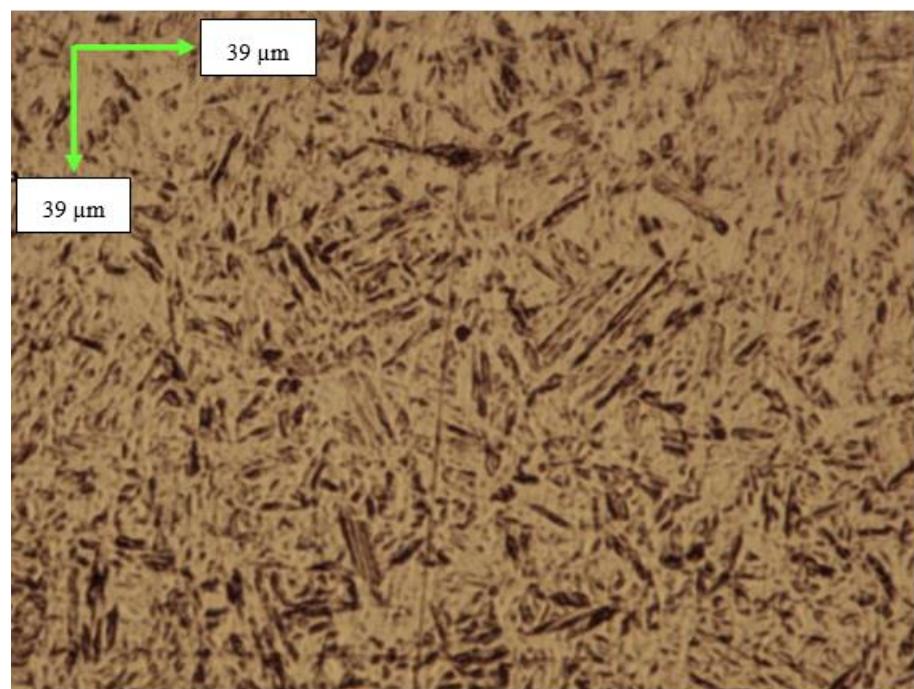


Figure 4. The microstructure for JIS S45C at 800 oC with 5 mesh size.

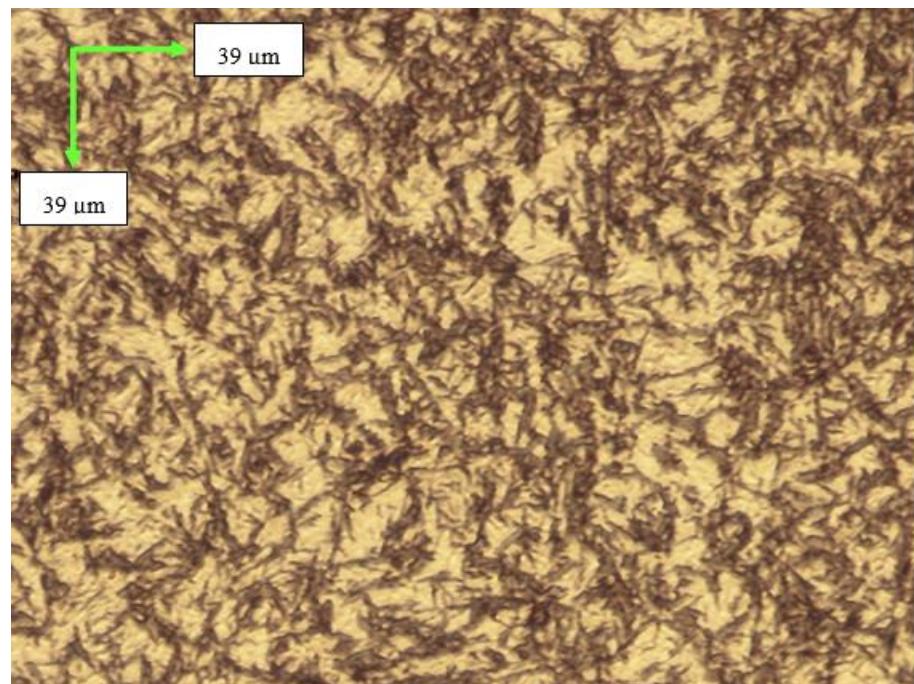


Figure 5. The microstructure for JIS S45C at 900 oC with 15 mesh size.

4. CONCLUSIONS

This preliminary study shows that the heating temperature and carbon grain size influenced the hardness and microstructure of JIS S45C steel. The higher heating temperature increased the hardness, and the smaller carbon grain size increased the hardness of the steel. However, further experiments are required to determine an optimum carbon grain size.

ACKNOWLEDGEMENTS

The authors would like to thank for the financial support provided by LPPM Sanata Dharma University.

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